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Practice Problem: Calculations Involving pH and Ka We know a bit

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about acids and bases, including the definitions of **pH and pKa**, as they relate to an acid-base equilibrium.

8.3 The pH scale (SL) 8.3 **pH and** change in hydrogen ion concentration
Understandings: **pH** = $-\log[\text{H}^+(\text{aq})]$
and $[\text{H}^+] = 10^{-\text{pH}}$. A change of one **pH** ...

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